

Exam. Code : 103203

Subject Code : 1304

B.A./B.Sc. 3<sup>rd</sup> Semester (Batch 2020-23)

### PHYSICS

#### Paper—A (Statistical Physics & Thermodynamics)

Time Allowed—3 Hours] [Maximum Marks—35

**Note** :— Attempt *five* questions in all, selecting at least *one* question from each section. **Fifth** question may be attempted from any section. Log tables can be asked for if necessary.

#### SECTION—A

1. Taking the case of  $n$  particles distributed in 2 compartments with equal a priori probability, discuss the variation of probability of a macrostate on account of small deviation from the state of maximum probability. 7
2. Four distinguishable particles are to be distributed among two compartments. The first compartment is divided into 3 cells and second into 2 cells. All the cells are of equal a priori probability and there is no restrictions on number of particles that can go into any cell. Calculate the values of  $W(4,0)$ ,  $W(3,1)$ ,  $W(2,2)$ ,  $W(1,3)$ ,  $W(0,4)$ . 7

#### SECTION—B

3. Treating ideal gas as a system governed by classical statistics, derive the Maxwell-Boltzmann law of distribution of molecular speeds. 7
4. Starting from the basic postulates, obtain the Fermi-Dirac distribution law. 7

#### SECTION—C

5. Discuss the thermodynamics of a thermocouple. Derive an expression for  $(dE/dT)$  and  $(d^2E/dT^2)$  for a thermocouple, where  $E$  and  $T$  have their usual meanings. 7
6. Derive an expression for the efficiency of the Carnot's heat engine, using one mole of an ideal gas as the working substance. 7

#### SECTION—D

7. (a) Derive an expression for  $(C_p - C_v)$  for van der Waal's gas. 5  
(b) Why does a rubber string heat up on stretching? 2
8. Starting from four thermodynamical potentials, derive Maxwell's thermodynamic relations. 7